## FERUGINIDIN AND FERUGIN, TWO NEW SESQUITERPENOIDS BASED ON THE CAROTANE SKELETON FROM FERULA JAESCHKEANA<sup>1</sup>

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About seventy species of the large genus *Ferula* (Umbelliferae) have already been investigated chemically. Most of the widespread sesquiterpenes of *Ferula* species are derivatives of monocyclic and bicyclic sesquiterpenes having the germacrane, humulane, carotane, himachalane, or guaiane skeleton (1-3).

In continuation of our studies on *Ferula jaeschkeana* Vatke (4), two new compounds tentatively named feruginidin [1] and ferugin [2] were isolated and characterized.

The methanolic extract of *F. jaesch*keana obtained after defatting the plant material with light petroleum ether (40-60°) and  $C_6H_6$  was chromatographed on silica gel with elution by CHCl<sub>3</sub> and CHCl<sub>3</sub>/MeOH (5 to 10%) mixtures which resulted in three gross fractions D, E, and F. Fraction F on repeated chromatography on silica gel column using light petroleum ether (40-60°)/ Me<sub>2</sub>CO (1 to 4%) mixtures followed by preparative tlc yielded two new compounds feruginidin [1] and ferugin [2].

Feruginidin [1], the  $5\alpha$ -*p*-hydroxybenzoyl ester of jaeschkeanadiol and ferugin [2], the  $5\alpha$ -*p*-hydroxybenzoyl ester of  $2\alpha$ -hydroxy-3(15)-ene jaesch-



<sup>1</sup>Central Institute of Medicinal and Aromatic Plants Publication No. 627. <sup>2</sup>Meerut University, Meerut, India. keanadiol, both analyzed for molecular formulas  $C_{22}H_{30}O_5$  and had mass spectra consistent with this composi-

tion. The uv maxima of both 1 and 2 (256 and 258 nm respectively), their ir bands at 3300-3350 (OH) and 1680 cm<sup>-1</sup> (ester), and their two pairs of two aromatic ortho-coupled protons indicated them to be an aromatic ester. Alkaline hydrolysis of 1 and 2 yielded phydroxybenzoic acid as one of the reaction products, confirming 1 and 2 as esters of this acid.

The ms of 1 and 2 showed a fragment at m/z 331 (M<sup>+</sup>-C<sub>3</sub>H<sub>7</sub>) arising from the loss of an isopropyl group which is a typical loss in carotane sesquiterpenes with a hydroxy function at C-7(5). The <sup>1</sup>Hnmr spectra (Table 1) of 1 and 2 were compared with a closely related sesquiterpene ester 3 (6) belonging to the carotane class of sesquiterpenes. Compound 1 had a signal for a hydroxymethyl group at  $\delta$  4.09 (br s) instead of for a vinylic methyl group at  $\delta$  1.84 (5) in 3. Compound 2 exhibited a secondary hydroxyl group at C-2 which is assigned the  $\alpha$ -stereochemistry on the basis of coupling constants of 6.0 Hz and 9.0 Hz between H-1 $\beta$  and H-2 $\beta$  and H-1 $\alpha$ , respectively. Compound 2 also showed signals for an exomethylene group at  $\delta$ 4.93, 5.23 (br s) instead of the vinylic methyl group in 3. The rest of the signals were similar to those of 3, and were assigned on the basis of chemical shifts and <sup>1</sup>H-nmr decouplings. The structure of 2 was further confirmed by comparing the <sup>1</sup>H-nmr spectrum of its methyl ether [4] with that of the corresponding reported reaction product of epoxyjaeschkeanadiol *p*-hydroxy benzoate (7).

## EXPERIMENTAL

Mps are uncorrected. Uv spectra were recorded in MeOH. <sup>1</sup>H-nmr spectra were taken at 400 MHz, TMS as internal standard. Eims and cims were obtained with a direct inlet system at 70 ev.

F. jaeschkeana rhizomes collected from Gulmarg (Kashmir) in November 1980 (voucher specimen No. 3013 preserved in the CIMAP herbarium), were air dried (1.6 kg), powdered, and

Proton Position	Compounds	
	Feruginidin (Hz)	Ferugin (Hz)
$ \begin{array}{c} H-1a & \dots & H-1a \\ H-1b & \dots & H-1b \\ H-2 & \dots & H-2a \\ H-4a & \dots & H-4a \\ H-5\alpha & \dots & H-5\alpha \\ H-6\alpha & \dots & H-8a \\ H-8b & \dots & H-8b \\ H-9a & \dots & H-9a \\ H-9a & \dots & H-9a \\ H-9a & \dots & H-8a \\ H-9a $	2. 17, dd, (15, 6) 1.57, dd, (15, 9) 5.82, dd, (9, 6) 2.48, dd, (15, 9) 2.04, dd, (15, 5) 5.20, ddd, (11, 9, 5) 2.04, d, (11) 1.25-1.59 m	2.15, dd, (12, 6) 1.30, dd, (12, 12) 4.18, dd, (12, 6) 2.64, dd, (15, 5) 2.57, dd, (15, 5) 5.38, ddd, (12, 10, 4) 2.13, d, (10) 1.3-1.65 m
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	1.94, qq, (7) 0.82, d, (7) 0.93, d, (7) 1.08, s 4.09, brs 	1.85, qq, (7) 0.80, d, (7) 0.89, d, (7) 1.18, s 5.23, brs 4.93, br s 6.82, dd, (9, 2) 7.85, dd, (9, 2) 7.85, dd, (9, 2) 6.82, dd, (9, 2)

TABLE 1. <sup>1</sup>H nmr of Feruginidin [1], Ferugin [2]<sup>a</sup>

\*The assignments have been made on the basis of chemical shifts and 1H decouplings.

extracted in a Soxhlet Apparatus with light petroleum ether (40-60°),  $C_6H_6$ , and MeOH, respectively. Solvents were removed under vacuum yielding three different fractions A, B, and C. The methanolic extract (fraction C), a viscous mass (115 g) was chromatographed on SiO<sub>2</sub> (2 kg) with CHCl<sub>3</sub>, CHCl<sub>3</sub>-MeOH (95:5), CHCl<sub>3</sub>-MeOH (90:10), and CHCl<sub>3</sub>-MeOH (1:1) successively. Concentrated CHCl<sub>3</sub>-MeOH (90:10) eluate (8.0 g) was further chromatographed on SiO<sub>2</sub> column with light petroleum ether (40-60°)-Me<sub>2</sub>CO (96:4) followed by preparative tlc separation in CHCl<sub>3</sub>-MeOH (85:15), yielding two compounds named as feruginidin (Rf 0.62) and ferugin (Rf 0.55).

Both gave a reddish color when sprayed with 1% solution of vanillin-H<sub>2</sub>SO<sub>4</sub> on SiO<sub>2</sub>-tlc plates on heating (10 min at 100°).

FERUGINIDIN.—Compound [1] (30 mg) colorless needles (MeOH), mp 130°;  $C_{22}H_{30}O_5$ (M+1, 375); uv  $\lambda$  max (MeOH) nm (log  $\epsilon$ ) 256 (4.12); ( $\alpha$ )<sup>20</sup>D + 6.0° (1.5, MeOH); ir (KBr)  $\nu$  max 3350, 2950, 2930, 2870, 1680, 1610, 1590, 1510, 1450, 1380, 1160, 1100, 1030, 850 cm<sup>-1</sup>; <sup>1</sup>H nmr see Table 1; cims *m*/*z* (CH<sub>4</sub>, % rel. abundance) 375 (M+1) (2), 357 (M+1-H<sub>2</sub>O)<sup>+</sup> (38), 331 (M-C<sub>3</sub>H<sub>7</sub>)<sup>+</sup>(1), 279 (14), 235 (20), 219 (58), 202 (23), 201 (M+1-2 H<sub>2</sub>O-C<sub>6</sub>H<sub>4</sub> (OH) COOH)<sup>+</sup> (100), 200 (32), 175 (3), 157 (4.5), 127 (27), 75 (6); (Found C, 70.9; H, 8.16; C<sub>22</sub>H<sub>30</sub>O<sub>5</sub> require C, 70.58; H, 8.02).

FERUGIN.—Compound [2] (25 mg) colorless needles (MeOH), mp 140°;  $C_{22}H_{30}O_5$  (356, M<sup>+</sup>- $H_2O$ ); uv  $\lambda$  max (MeOH) nm (log  $\epsilon$ ) 258 (3.98); [ $\alpha$ ]<sup>20</sup>D + 15.0° (1.5, MeOH); ir (KBr)  $\nu$  max 3300, 2980, 2920, 2880, 1680, 1610, 1590, 1460, 1410, 1370, 1280, 1160, 1120, 1050, 920, 850 cm<sup>-1</sup>; ms m/z (% rel. abundance) 356 (M-H<sub>2</sub>O)<sup>+</sup> (20), 338 (M-2H<sub>2</sub>O)<sup>+</sup> (5), 331 (M- $C_3H_7$ )<sup>+</sup> (6.5), 330 (40), 312 (40), 236 (M-C<sub>6</sub>H<sub>4</sub> (OH) COOH-<sup>+</sup> (5.5), 218 (4.5), 200 (20), 193 (M-C<sub>6</sub>H<sub>4</sub> (OH) COOH-C<sub>3</sub>H<sub>7</sub>)<sup>+</sup> (12), 175 (45), 157 (8), 132 (30), 121 (C<sub>6</sub>H<sub>4</sub> (OH) COOH-OH)<sup>+</sup> (100), 93 (25), 83 (5), 77 (5), 71 (25), 65 (70) (Found C, 70.80; H, 8.10; C<sub>22</sub>H<sub>30</sub>O<sub>5</sub> require C, 70.58; H, 8.02). Treatment of 2 with ethereal  $CH_2N_2$  afforded 4 identical with the product reported by Fraga *et al.* (7).

HYDROLYSIS OF FERUGINIDIN [1] AND FERUGIN [2].—A mixture of 1 (10 mg) and 4% methanolic aqueous KOH (2 ml) was refluxed for 2 h. and worked up for the acidic fraction by the usual method.

The acidic fraction Rf 0.85 ( $C_6H_6$ -MeOH-HOAc, 90:16:8:) showed one spot identical with an authentic sample of *p*-hydroxybenzoic acid, mp 210-214°. Similar hydrolysis of **2** also gave *p*-hydroxybenzoic acid.

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